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- (54) Cellulose powder and method of producing the same.
- (57) A fine cellulose powder consists of rounded particles, has a viscosity average polymerization degree of 20 150, and exhibits a water absorption coefficient of 100 % or more relative to its own weight. A method of producing the fine cellulose powder comprises the steps of aging alkali cellulose to reduce its average polymerization degree and increase its friability, neutralizing the aged alkali cellulose, and pulverizing the result, preferably in a jet mill.

## Cellulose powder and method of producing the same.

Description	of	EP	051	520	8
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Translate this text

This invention relates to a cellulose powder. A preferred embodiment consists of rounded particles and may exhibit a water absorption coefficient of not less than 100% relative to its own weight. The viscosity average polymerization degree (hereinafter referred to simply as "average polymerization degree") is 20-150. A method of producing the same is also provided.

A number of methods are known for obtaining fine cellulose powder. These include:

(1) A method of producing fine crystalline cellulose by acid hydrolyzation of the non - crystalline portions of cellulose of wood, cotton or the like and pulverizing the result. However, this method has shortcomings in that the powder particles are angular owing to the presence of residual crystalline portions in the hydrolyzed and removed non - crystalline portions, and also in that the product powder is expensive and exhibits poor water absorption

(2) A method in which cellulose or a derivative thereof is coagulated by dripping through capillaries into a liquid coagulant. This method has the drawbacks of low productivity, difficulty of producing fine particles and

very high operating cost.

(3) A method in which a solution of acetyl cellulose or other cellulose compound soluble in an organic solvent is spray - dried and then converted back into cellulose by deacetylization or the like. This method also has various disadvantages, such that it involves complicated processing steps and requires recovery of the solvent.

This invention was accomplished in view of the fact that commercial production of fine cellulose powder exhibiting high water absorption is difficult to achieve by prior art methods because of the complexity of the production process, the difficulty encountered in the handling of intermediate raw materials and chemicals, and the high cost of production. A preferred embodiment of the invention may provide a fine cellulose powder consisting of rounded particles and exhibiting a high water absorption, and a simple and inexpensive method of producing the same.

The fine cellulose powder according to the present invention exhibits an average polymerization degree of 20 - 150 as measured according to the method stipulated by JIS (Japanese Industrial Standard) L1015 - 1981. Up to now, it has not been possible to achieve a high water absorption coefficient with a cellulose powder having a polymerization degree of such a low level. The particle diameter of the fine cellulose powder is small, specifically between 0.1 mu m and 50 mu m. The swelling factor of the fine powder in water is desirably not less than 2 based on the particle volume of the powder in dry state taken as 1.

The water absorption coefficient of the fine powder measured by soaking it in water for 30 minutes under reduced pressure and then freeing it of non-absorbed water by centrifuging at 3000 rpm for 30 minutes is preferably not less than 100% relative to the dry weight of the fine powder.

The water absorption coefficient referred to here is defined as  $(W - w) / w \times 100(\%)$ 

where w is the weight of the fine powder before water absorption and W is the weight thereof after water absorption.

The fine powder of this invention can be produced by neutralizing brittle alkali cellulose whose average polymerization degree has been reduced by aging and pulverizing the result.

The polymerization degree of alkali cellulose is reduced by prolonged aging and this facilitates pulverization after neutralization and drying.

Alkali cellulose is prepared for use as a raw material. While the conditions for preparing the alkali cellulose are not particularly defined, it is typically prepared by soaking cellulose in a 14 DEG C, 17.5 % caustic soda solution and compressing the result to increase its specific weight by a facter of 2.8, followed by pulverization in a kneader or other such grinding machine.

The aging may be conducted at a relative humidity of not less than 70 %, preferably not less than 90 %,

and not more than 100 % and a temperature of not less than 30 DEG C and not more than 100 DEG C. preferably not less than 45 DEG C and not more than 75 DEG C, for not less than 1 day and not more than 20 days. This aging provides an alkali cellulose exhibiting an average polymerization degree of around 30 -200.

The aged alkali cellulose may be neutralized by casting into an aqueous solution containing at least a stoichiometric amount of sulfuric acid, hydrochloric acid or the like relative to the alkali content of the alkali cellulose.

The Product of this neutralization may be dried in an ordinary constant - temperature drying oven or the like, and then pulverized. Pulverization can be conducted using a ball mill or other ordinary type of dry grinding machine. The use of a jet mill is particularly advantageous. This pulverization yields a fine cellulose powder having an average polymerization degree of 20 - 150.

While the particles of the fine powder obtained according to this invention may have differing shapes they are invariably rounded. Although the particle diameter varies with the production conditions, production of a fine powder with a particle diameter falling in the range of 0.1 - 50 mu m can be readily achieved. The swelling factor of the fine powder in water is not less than 2.

## DESCRIPTION OF THE PREFERRED EMBODIMENTS

The invention will now be explained with reference to specific examples.

### Example 1

Pulp soaked in a 17.5 wt % aqueous solution of sodium hydroxide at 14 DEG C for 1 hour was squeezed to 2.8 times its original specific gravity and left to stand for 5 days at 50 DEG C.

The resulting alkali cellulose, which had an average polymerization degree of 120, was neutralized in a 5 %aqueous sulfuric acid solution under thorough stirring. The result was thoroughly washed with water, filtered to obtain a wet cake, completely broken up, dried in a constant - temperature drying oven, and then pulverized in a jet mill.

The diameters of particles of the cellulose powder obtained in this way fell in the range of 0.4 - 30 mu m and averaged 6.0 mu m in dry state. In water they fell in the range of 1 - 80 mu m and averaged 14.0 mu m. The water absorption coefficient of the powder was 160 %. Its average polymerization degree as measured in accordance with JIS L - 1015 - 1981 was 110.

The water absorption coefficient of the powder measured by soaking it in water for 30 minutes under reduced pressure and then freeing it of non - absorbed water by centrifuging at 3000 rpm for 30 minutes was 160 % relative to its own weight.

#### Example 2

Pulp soaked in a 17.5 wt % aqueous solution of sodium hydroxide at 14 DEG C for 1 hour was squeezed to 2.8 times its original specific gravity and left to stand for 10 days at 50 DEG C.

The resulting alkali cellulose, which had an average polymerization degree of 90, was neutralized in a 5 % aqueous sulfuric acid solution under thorough stirring. The result was thoroughly washed with water, filtered to obtain a wet cake, completely broken up, dried in a constant - temperature drying oven, and then pulverized in a jet mill.

The diameters of the particles of the cellulose powder obtained in this way fell in the range of 0.1 - 7 mu m and averaged 4.0 mu m in dry state. In water they fell in the range of 0.3 - 20 mu m and average 10.0 mu m. The water absorption coefficient of the powder was 180%. Its average polymerization degree as

measured in accordance with JIS L - 1015 - 1981 was 70.

### Comparative example

Commercially available cellulose powder produced by hydrolyzing the non - crystalline portions of pulp but still containing crystalline portions was tested to determine its properties. The diameters of its particles were found to fall in the range of 5 - 80 mu m and average 30.0 mu m in dry state and to fall in the range of 5 - 80 mu m and average 31.5 mu m in water. Its water absorption coefficient was measured to be 98%.

The fine cellulose powder according to the invention has excellent swelling property and exhibits good absorption property with respect to a wide range of liquids, including aromatization agents, pesticides, antiseptics, dye solvents and ink. It can thus be used as a carrier for these liquides or as a filler for various purposes.

The conventional fine cellulose powder produced by hydrolyzing the non - crystalline portions of pulp or the like but still containing crystalline portions has a large particle size and exhibits inadequate absorption property with respect to various substances. In contrast, the fine cellulose powder according to this invention consists of particles that are not only small in diameter but also rounded, and further exhibits absorption property with respect to various substances. It is thus extremely useful.

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# Cellulose powder and method of producing the same.

Claims of **EP0515208** 

Translate this text

- 1. A cellulose powder having a viscosity average polymerization degree of 20 150 and a particle diameter of 0.1 - 50 mu m,
- 2. A cellulose powder according to claim 1 having a swelling factor in water of not less than 2.
- 3: A cellulose powder according to claim 1 having a water absorption coefficient of not less than 100 % as measured by soaking in water for 30 minutes under reduced pressure followed by centrifuging at 3000 rpm for 30 minutes for removal of non - absorbed water.
- 4. A method of producing fine cellulose powder comprising the steps of neutralizing brittle alkali cellulose whose average polymerization degree has been reduced by aging and pulverizing the result.

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### **EUROPEAN SEARCH REPORT**

Application Number

92 30 4665

Category	Citation of document with in of relevant pas		Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl.5)	
х	GB-A-663 621 (ANGLO INDUSTRIES LIMITED)  * page 1, line 52 -  * page 2, line 22 -  * page 3, line 33 -	line 76 * line 26 *	4	C08B15/02 C08B1/08 C08J3/12 //C08L1/02	
X	CHEMICAL ABSTRACTS, 1972, Columbus, Ohio abstract no. 17867, microcrystalline ce cellulose' * abstract * & KOBUNSHI KAGAKU vol. 29, no. 9, 1972 pages 647 - 651 ARAKI T. ET AL.	o, US; Preparation of Hulose by aging alkali	4		
X	WO-A-8 000 843 (BATE INSTITUTE) * claims 1-3; examp		1		
A	EP-A-0 200 973 (KANE * abstract; claims 4		1-3	TECHNICAL FIELDS SEARCHED (Int. Cl.5)	
Ρ,χ		SHINBO INDUSTRIES INC.) - line 15; example 1 *		CO8J	
<b>A</b>	US-A-3 397 198 (D. 5 * column 1, line 10 * column 2, line 15 * column 2, line 28 * column 2, line 67	- line 16 * - line 18 * - line 30 *	1-3		
<b>A</b>	INDUSTRIAL AND ENGING vol. 54, no. 9, 1962 pages 20 - 29 O. A. BATTISTA ET Al cellulose'	2,			
	The present search report has be	en drawn up for all claims			
•	Place of search THE HAGUE	Date of completion of the search 23 NOVEMBER 1992		MAZET J.	
X : par Y : par doc	CATEGORY OF CITED DOCUMEN ticularly relevant if taken alons ticularly relevant if combined with ano ument of the same category hadological background	E : earlier patent d after the filing ther D : document cited L : document cited	ocument, but put date in the applicatio for other reasons	dished on, or n	

# EP0515208 - Cellulose powder and method of producing the same - NISSHINBO INDUSTRIES, INC.

# Database down

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## EP0515208 - Cellulose powder and method of producing the same

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For all designated states

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[1992/48]

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Title

German

Zellulosepulver und Verfahren zu seiner

Herstellung [1992/48]

English

Cellulose powder and method of producing the

same [1992/48]

French

Poudre de celluiose et méthode pour sa

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Penalty fee

Penalty fee Rule 85b

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M01 Not paid yet

**Documents cited:** 

Search

[X] GB663621 💎

[X] WO8000843 **\***[A] EP0200973 **\***[XP] EP0460609 **\***[A] US3397198 **\*** 

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